

**Design of Question Paper
Science (086),
Summative Assessment-II
Class X – (2012-13)**

WEIGHTAGE

S. No.	Name of the unit	Chapter name	Weightage
1.	Chemical substances Nature and Behaviour	i) Carbon Compounds ii) Periodic classification of elements	23
2.	World of Living	i) How do organisms reproduce ii) Heredity and evolution	30
3.	Nature Phenomena	i) Light - reflection and refraction ii) Human eye and the colourful world	29
4.	Natural resources	i) Our environment ii) Management of Natural resources	8

**The question paper will include value based question(s) to
the extent of 3-5 marks.**

Types of questions	Marks per question	Total no of questions	Total marks
MCQ	1	18	18
VSA	1	3	3
SA (I)	2	4	8
SA (II)	3	12	36
LA	5	5	25
Total		42	90

**Sample Questions
Science (086),
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MCQ

Que. 1 Acetic acid was added to four test tubes containing the following chemicals:

- a. Sodium carbonate
- b. Blue litmus solution
- c. Lime water
- d. Distilled water

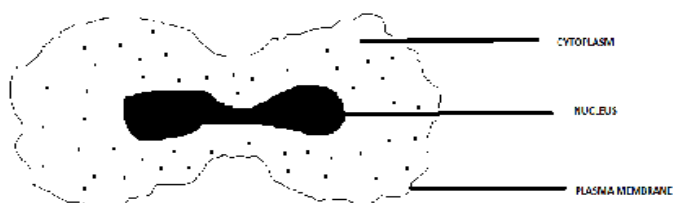
Which amongst these is/ are correct option(s) for carrying out a characteristic test for identification of a carboxylic acid (acetic acid) in the laboratory?

1. (a) only
2. (c) only
3. (a) and (b)
4. (c) and (d)

Que. 2 On adding concentrated NaOH solution to a test tube containing phenolphthalein, the colour change observed by a student would be:

- a. Pink to colourless
- b. Pink to blue
- c. Colourless to pink
- d. Red to blue

Que. 3 A student observed a focussed slide of a stage of binary fission in Amoeba under a microscope as seen below:



Which of the following is most appropriate statement with respect to the observation-

- a) preliminary stage
- b) intermediate stage where the nucleus is divided and cytoplasm is not
- c) intermediate stage where the cytoplasm is divided and nucleus is not
- d) final stage

Que. 4 A student while observing the properties of acetic acid would report that this acid smells like

- (i) vinegar and turns red litmus blue
- (ii) rotten egg and turns red litmus blue
- (iii) vinegar and turns blue litmus red
- (iv) rotten egg and turns blue litmus red

VSA

Que. 5 Give one example of a unisexual flower.

Que. 6 Draw the electron dot structure of the gas molecule which is liberated when zinc metal is treated with aqueous NaOH solution.

Que. 7 Find the period and group of the element whose atomic number is 12.

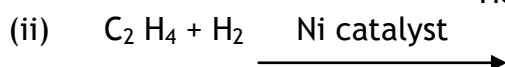
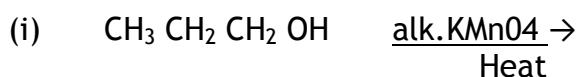
Que. 8 Refractive index of two material medium X and Y are 1.3 and 1.5 respectively. In which of the two, the light would travel faster?

Que. 9 Write in one word or at the most in one sentence about the following

- (i) Mirrors used by dentists to examine teeth
- (ii) The smallest distance, at which the eye can see objects clearly without strain.

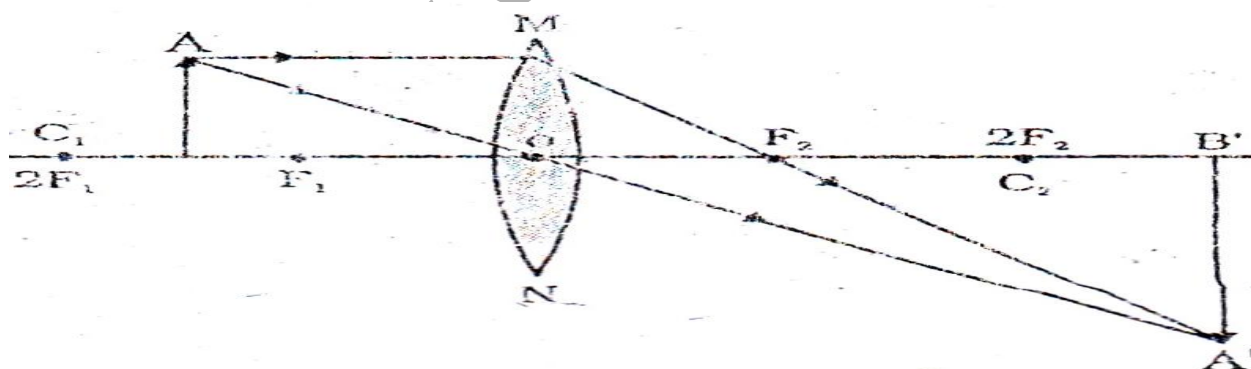
SA I

- Que. 10 List two differences between acquired and inherited traits.
- Que. 11 State how would you distinguish between Acetic acid and Ethanol in your laboratory. Give chemical equation of the reactions shown by them. Write the chemical equations involved.
- Que. 12 Complete the reaction(s) given below and classify them as Combustion / Oxidation / Addition / Substitution reaction.



- Que. 13 Explain why?
- (i) 'Danger' signal are red in colour.
- (ii) Convex mirrors are commonly used as rear - view mirrors.

- Que. 14 Study the ray diagram given below and answer the following questions -



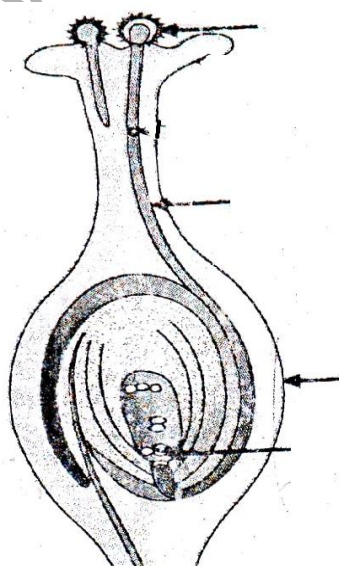
- (i) State the type of lens used in the figure.
- (ii) List two properties of the image formed.
- (iii) In which position of the object will the magnification be -1?
- Que. 15 What is meant by power of accommodation of the eye? How is it related to the focal length of the eye lens?

SA II

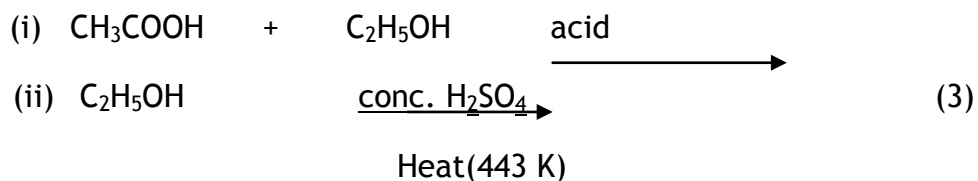
- Que. 16 “Fossils are related to evolution”, justify this statement. Give the two ways by which age of the fossils can be estimated?
- Que. 17 ‘Variation is beneficial to the species but not necessarily for the individual’, give three reasons to justify it.
- Que. 18 Two elements with symbol X (atomic no. 11) and Y (atomic no. 13) are placed in the III period of the modern periodic table -
- Which amongst the two has more metallic character?
 - Calculate the valency of each element.
 - Element ‘Y’ is smaller than ‘X’ in terms of atomic size. Is the statement true, justify?

LA

- Que. 19
- Identify the parts the 1, 2, 3 and 4 of the diagram given below.
 - List two changes that can be observed in the flower after fertilization.



Que. 20 a) Complete the following reactions / chemical equations and name the main product formed in each case-



Also state the importance of conc. H_2SO_4 in this reaction.

b) List two reasons why carbon forms large number of compounds and are poor conductors of electricity? (2)

Que. 21 A 2.0 cm tall object is placed perpendicular to the principal axis of a concave lens of focal length 10 cm. The distance of the object from the mirror is 15 cm. Find the nature, position and size of the image formed. Represent the situation with the help of a ray diagram.

MARKING SCHEME

MCQ

Que. 1	(3)	1
Que. 2	(c)	1
Que. 3	(b)	1
Que. 4	(i)	1

VSA

Que. 5	Watermelon / Papaya/ Cucurbita pepo	1
Que. 6	Hydrogen gas, H : H	1
Que. 7	3 rd period, 2 nd group	1
Que. 8	in medium 'X' because of lower value of refractive index	1
Que. 9	(i) Concave Mirror	1
	(ii) Near point	1

SAI

Que. 10	Acquired	Inherited	
	<ul style="list-style-type: none"> Are not passed on to next Generation They cannot direct Evolution 	<ul style="list-style-type: none"> Can be passed on to next generation They bring about evolution of species 	$\left. \begin{array}{l} \frac{1}{2}, \frac{1}{2} \\ \frac{1}{2}, \frac{1}{2} \end{array} \right\} 2$

Que. 11	<p>Activity:</p> <p>Take sodium carbonate in two test tubes.</p> <p>Mark them as 1 and 2.</p> <p>Pour some amount of acetic acid in test tube 1 and ethanol in 2.</p> <p>Effervescence produced in test tube 1 confirms production of gas which can not be found in test tube 2.</p> <p>Conclude that the reaction is taking place in with acetic acid and not with ethanol.</p> $2\text{CH}_3\text{COOH} + \text{Na}_2\text{CO}_3 \longrightarrow 2\text{CH}_3\text{COONa} + \text{H}_2\text{O} + \text{CO}_2$	1
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	$C_2H_5OH + Na_2CO_3 \longrightarrow$ No reaction	1	
Que. 12	(i) CH_3CH_2COOH , Oxidation (ii) C_2H_6 , Addition		2
Que. 13	(i) Does not scatter - red light intense	2	
	(ii) Wider view		
Que. 14	(i) Convex lens	2	
	(ii) Real, inverted, enlarged (any two)		
	(iii) When object is at 2F		
Que. 15	The ability of the eye to adjust its focal length.	1	
	Change in curvature of the eye lens can thus change its		1
	Focal length, muscle relaxed, eye lens becomes thin or thick which helps to see objects clearly.		
SA II			
Que. 16	Preserved traces of living organisms are called fossils	1	} 3
	Two ways of determining age of fossils		
	Relative - fossils closer to the surface are more recent	1	
	Dating - finding the ratio of different isotopes	1	
Que. 17	For existence of the individual, variation may not be of importance but for the existence of a species it is, because -		3
	It brings in the resistivity wherever required. (survival)		
	It is responsible for acclimatization in varied environmental conditions. (adjustment)		
	It also makes one species different from the other. (diversity)		
Que. 18	i) Y		
	(ii) 2 each		
	(iii) false, same period - y lies in group 2 and x in group 16		

LA

- Que. 19 a) 1) Pollen grain
2) Pollen tube
3) Ovary
4) Female germ cell
- 5
} $\frac{1}{2} \times 4$

- b) Ovule changes into seed, ovary ripens and makes fruit, petals, sepals,
Stamens, style shrivel and fall off 3

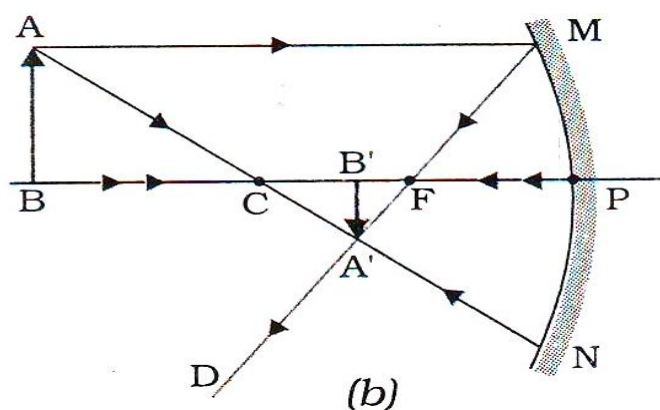
- Que. 20 (a) (i) $\text{CH}_3\text{COOC}_2\text{H}_5$ (Ethyl Ethanoate) (1+1)
(ii) CH_3COOH (Ethanoic acid), conc. H_2SO_4 absorbs water produced in
the reaction which may otherwise reverse the reaction (1)

- (b) Catenation, covalent compounds (1+1)

- Que. 21 $(1/v) + (1/u) = (1/f)$, $u = -15\text{cm}$, $f = -10\text{cm}$

$$(1/f) - (1/u) = (1/v)$$

$$\frac{1}{-10} - \frac{1}{-15} = \frac{1}{v} = V = -30 \text{ cm}$$



Value Based Questions
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Question:

Ethanol, commonly called as alcohol is an excellent solvent, is used in medicines and is an important chemical compound involved in synthesis of many chemical compounds. However in spite of its benefits to man, its impact on social behaviour has always been questioned. Media has often show abnormal behaviour of people while drunk. It is considered as a curse in the lives of those who are addicted to alcohol - 'Alcoholic' people are not only lowering their metabolism and affecting Central Nervous System, they are also a threat to the lives of others. Anger and rude behaviour are some of its ill effects.

- (i) Comment on the statement - 'Should production of alcohol should be banned', give three valid reasons to justify.
- (ii) As a student what initiative would you take in the common concern of 'Save Life, Do not Drink'. Give two suggestions.

Answer:

In favour of negative response: (3)

- (i) regulate production and supply
- (ii) it is used in so many ways for medicines, ornamentation
- (iii) it is used as disinfectant

In Favour of positive response:

- (i) cause of death of many people
- (ii) many adolescents get affected out of it and become addict
- (iii) is being misused even where it is of important use (for example, painting shops, industries)

Initiatives: (2)

Drive to make aware
Skit / Role plays / drama
Article writing
Chart preparation
Slogan writing

Question:

'Sania and Shreya' are best friends and study in grade 4, recently, Sania has been facing difficulty in reading the black-board text from the last desk. Shreya is little uncomfortable and wonders why sania avoids sitting on the last desk. On observation she found that sania often carries junk food in her lunch. Shreya has started sharing her lunch - full of green vegetables and fruits with her. Sania is now better and has also started taking a 'balanced diet'.

- (i) Name the eye defect Sania is suffering from?
- (ii) What are the two possible deformities related to her eye defect?
- (iii) What value is shown by Shreya and Sania?

Answers:

- (i) Myopia, short sightedness 1
- (ii) Lens defect (increased thinness), eye ball defect (shortening) 2
- (iii) Friendship, concern for each other, value and balanced diet 2